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A battery, which is actually an electric cell, is a device that produces electricity from a chemical reaction. Strictly speaking, a battery consists of two or more cells connected in series or parallel, but the term is generally used for a single cell. A cell consists of a negative electrode; an electrolyte, which conducts ions; a separator, also an ion conductor; and a positive electrode. **Continue with How Batteries Work/Battery Types**

The History of Batteries

Alessandro Volta experimented with zinc and silver plates to produce electric currents. Volta stacked the two to form a "pile", the first "dry" battery. By 1866, Georges Leclanche, a French engineer, patented a new system, which was immediately successful. Leclanche's "wet" cell (as it was popularly referred to) became the forerunner to the world's first widely used battery, the zinc carbon cell.

History of Batteries

Batteries are one of the most important objects in our everyday lives. Learn about the history of batteries, the basic concept of the operation of batteries and new advancements in batteries.

Battery

In 1999 we celebrated the 200th anniversary of the invention of the battery by Alessandro Volta.

Alessandro Volta

Biography of the inventor of the battery, Alessandro Volta.

Alkaline Battery

Lew Urry developed the small alkaline battery in 1949. The inventor was working for the

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